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Unexpected Entry to a Novel Carbocyclic Tricyclic System

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The bis(3,4-dimethyoxyphenylcyclopropyl) ketone (1b) undergoes rapid rearrangement in the presence of stannic chloride in nitromethane to the new highly symmetrical dibenzotricyclic systems (3).

Previously we have described the Lewis acid catalysed rearrangement of arylcyclopropyl ketones to α -tetralones¹ and developed it into a methodology for the synthesis of picropophyllone² and epipodophyllotoxin.³ A related approach has been successfully applied to the synthesis of steganone by Magnus and co-workers.⁴ Subsequently, we undertook an

investigation of bis(cyclopropyl)ketones. We noted that bis(p-methoxyphenylcyclopropyl) ketone (1a) also undergoes rapid rearrangement under the same conditions [SnCl₄, (0.06 M, 1.5 equiv.), MeNO₂, room temp., 20 min]. An epimeric mixture of benzindenones, (2), is formed.⁵

We now find that when the aryl groups in (1) are further



activated as in (1b) the reaction follows a new path.* The first example of the new dibenzo tricyclic system (3) \dagger , \ddagger is formed. The structure of this highly symmetrical, relatively strain-free compound has been unambiguously determined by spectroscopic methods.* Although the detailed reaction pathway of the conversion of (1b) to (3) remains to be determined, it appears that a combination of the activating effect of the second aryl methoxyl group in (1b), coupled with the requisite *endo* stereochemistry of α -(2b), is crucial to the success of this

 ${}^{+}$ C₂₃H₂₆O₄, *M*⁺, 366; 14% yield; m.p. 151–152 °C; δ_H(400 MHz, CDCl₃ with additional comprehensive decoupling experiments at 250 MHz) 6.7 (s, 2 × ArH), 6.64 (s, 2 × ArH), 3.87 (6 H, s, 2 × OMe), 3.81 (6 H, s, 2 × OMe) 3.71 (dt, J_{1,2} 3.2 and J_{1,3} 1.6 Hz, 1-H), 3.34 (2 H, m, 4-H and 7-H), 2.75 (1 H, dtt J_{1,3} 1.6, J_{2,3} 3.2, and J_{3,4} 8.0 Hz, 3-H), 2.11 (2 H, m, J_{58,5α} 7.2, J_{58,6α} 4.8, J_{4,58} 8.0 Hz, 5β-H and 6β-H), 2.07 (2 H, t, J_{3,4} 8.0, J_{4,5} 8.0 Hz, 4-H and 7-H), 1.51 (2 H, m, J_{58,5α} 7.2, J_{5α,6β} 4.8, J_{4,5α} 8.0 Hz, 5α-H and 6β-H), 2.07 (2 H, t, J_{3,4} 8.0, J_{4,5} 8.0 Hz, 5α-H and 6β-H), 2.07 (2 H, t, J_{3,4} 8.0, J_{4,5} 8.0 Hz, 5α-H and 6β-H), 2.07 (2 H, t, J_{3,4} 8.0, J_{4,5} 8.0 Hz, 5α-H and 6β-H), 2.07 (2 H, t, J_{3,4} 8.0, J_{4,5} 8.0 Hz, 5α-H and 6β-H), 2.07 (2 H, t, J_{3,4} 8.0, J_{4,5} 8.0 Hz, 5α-H and 6β-H), 2.07 (2 H, t, J_{3,4} 8.0, J_{4,5} 8.0 Hz, 5α-H and 6β-H), 2.07 (2 H, t, J_{3,4} 8.0, J_{4,5} 8.0 Hz, 5α-H and 6β-H), 2.07 (2 H, t, J_{3,4} 8.0, J_{4,5} 8.0 Hz, 5α-H and 6β-H), 2.07 (2 H, t, J_{3,4} 8.0, J_{4,5} 8.0 Hz, 5α-H and 6α-H); δ_c(62.5 Mz, assignments confirmed by a series of INEPT⁶ spectra) 148.84 (s, C-5'), 147.81 (s, C(4'), 134.43 (s, C-1'), 132.22 (s, C-2'), 111.56 (d, C-3'), 110.65 (d, C-6'), 56.01 (q, Me), 40.35 (d, C-1), 42.23 (d, C-4, C-7), 36.58 (d, C-3), 35.15 (t, C-6 and C-5) and 28.98 (t, C-2).

conversion. Consistent with this hypothesis, was the isolation of β -(2b) from the reaction of (1b). None of α -(2b) was detected. We consider that (5) slowly forms a carbocation which by an intermolecular hydride abstraction undergoes conversion to (3). Consistently, the dehydro product (4) was also formed in the reaction.

Although (3) is formed in low yield, this one-step synthesis from readily accessible ⁵ (1b) * opens for study this interesting new ring system. The analogous product * is formed in similar yield when the substrate (1) $(R^{1}-R^{2} = -OCH_{2}O-)$ is employed.⁷

References

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^{*} The conditions employed were: $SnCl_4$ (0.1M, 1.5 equiv.), $MeNO_2$, room temp., 5 min. The substrates (1a) and (1b) thus appear to have similar reactivities.

[†] All new compounds had satisfactory microanalysis and unequivocal spectra.